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p Postulates of
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atomic mass of first

20 elements of the

periodic table *Modern*

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Law of Multiple

Proportions (Filipino-

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Atomic Model

(Chapter 3, Class IX ,
N.C.E.R.T) *Section 3*

Modern Atomic

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Theory Atomic Theory
through the
Nineteenth Century.

The earliest recorded
discussion of the
basic structure of
matter comes...

Radioactivity. The

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second line of
investigation began in
1896, when the
French physicist Henri
Becquerel
(1852–1908)... The
Atomic Model. Once
scientists ...

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continued GROUND
STATE AND

EXCITED STATE The

lowest state of energy

of an electron is

called its ground

state. When an

electron gains energy,

it moves to an excited

state in a higher

energy level.

Electrons gain energy

by absorbing photons.

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A photon is the smallest unit of light energy.

CHAPTER 4 Atoms
SECTION 3 Modern Atomic Theory

Modern Models of the Atom. In the modern atomic model, electrons can be found only in certain energy levels, not between levels.

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Electron location is limited to energy levels. Electrons act like waves. The exact location of an electron cannot be determined. Orbital: a region in an atom where there is a high probability of finding electrons

*Section 3: Modern
Atomic Theory*

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Eventually, you will

agree to discover a

further experience

and carrying out by

spending more cash.

yet when? complete

you say yes that you

require to get those all

needs following

having significantly

cash? Why don't you

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attempt to acquire
something basic in
the beginning?

Section 3 Modern

Atomic Theory

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Theory and Electrons

Chapter 4 Section 3 .

Energy Levels ! In

Bohr's model,

electrons move at a ...

Atomic Orbitals ! The

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Section 3

electron cloud

represents all the ... 3!

9! 18! 4! 16! 32!

Electron

Configurations !

Modern Atomic

Theory and Electrons

Matter is composed of

exceedingly small

particles called atoms.

An atom is the

smallest unit of an

element that can

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participate in a chemical change. An element consists of only one type of atom, which has a mass that is characteristic of the element and is the same for all atoms of that element (Figure 2.3. 1).

2.3: Modern Atomic Theory and the Laws That Led to It ...

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3—3 Review and

Reinforcement

Modern Atomic

Theory Complete the

following table. 1. 2.

3. If the statement is

true, write “true.” If it

is false, change the

underlined word or

words to make the

statement true. Write

your answer on the

line provided. 4.

Moseley discovered

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that all atoms of an element have the same number of neutrons in ...

Answers

Name Date Class 3—3

Review and

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Theory

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and motion. (Section

3.1) Describe the

nuclear model of the

atom. (Section 3.4)

Describe the

similarities and

differences between

solids, liquids, and

gases with reference

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Section 3

to the particle nature of matter, the degree of motion of the particles, and the degree of attraction between the particles. (Section 3.1) Review Skills

Chapter Modern a theory

What did Bohr contribute to modern atomic theory? Bohr

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Section 3

contributed a
description of the
arrangement of
electrons in an atom.

... Science 4.3

Chapter 4 section 3.

16 terms. gonzalis. 04

- Atomic Structure:

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PosberghTenafly.

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5 terms. jason8876.

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the elements . Atomic
Spectra . Bohr model
of the atom . Dalton's
atomic theory quiz.

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[Arrangement of
Electrons in Atoms]

Chapter 4 :

Arrangement of

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electrons in atoms.

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Modern Chemistry by
Holt, Rinehart, and
Winston on Chapters
4 and 5, which deals
with electrons and the
periodic table.

Chapter 4

Arrangement Of

Electrons In Atoms

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Chemistry at

Thornapple Kellogg

High School.

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Assessment Chapter

4: Practice Quiz

Section 3: Modern

Atomic Theory In

Modern Atomic

Theory Practice Quiz

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Theory

Chemistry is based on the modern atomic theory, which states that all matter is composed of atoms.

Atoms themselves are composed of protons, neutrons, and electrons. Each element has its own atomic number, which is equal to the number

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Section 3

of protons in its nucleus. Isotopes of an element contain different numbers of neutrons.

3.1 Atomic Theory - lardbucket

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Pied Danser Faux
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Chapter 4 Atomic
Structure Section 4.3
Modern Atomic
Theory (pages
113118) . Use this
table to answer
questions 12 and 13..
Name Chapter 4:
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We will investigate
that more = in the
next Section

SECTION 3-1 HW:
Page 69; 1, 2, & = 3
and Page 87; 1 & 2 .
Page 69. 1. Describe
the major
contributions of each
of the following to the

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