

Chemistry Practical And Questions Answers 2014

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Chemistry 3 (Practical) (Alternative A) 12.00pm - 2.00pm (2nd Set) 1. A is 0.100 mol dm⁻³ solution of an acid. B is a solution KOH containing 2.8 g per 500 cm³. (a) Put A into the burette and titrate it against 25.0 cm³ portions B using methyl orange as indicator. Repeat the titration to obtain consistent titres.

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D) artificial radioactivity. Answer & Explanation Answer: B) nuclear fusion. Explanation: Hydrogen Bomb is based on the principle of Nuclear Fusion. Nuclear fusion is a reaction in which two or more atomic nuclei are combined to form one or more different atomic nuclei and subatomic particles.

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Guide to Chemistry Practicals - TETEA

A solution containing one gram-equivalent mass of the solute per litre of the solution is called a normal solution. Question.3.What is the equivalent mass of KMnO_4 when it acts as oxidizing agent in acidic medium ? Answer. KMnO_4 loses 5 electrons per molecule, when it acts as oxidizing agent in the presence of acids.

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Titration questions. Google Classroom Facebook Twitter. Email. Titrations . Practice: Titration questions. This is the currently selected item. Acid-base titrations. Worked example: Determining solute concentration by acid-base titration. Titration of a strong acid with a strong base.

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Density Test Questions with Answers

An important laboratory technique in a chemistry lab is the ability to accurately measure a liquid in a graduated cylinder. This is a collection of questions dealing with reading the meniscus of a liquid. Remember that the meniscus is the curve seen at the top of a liquid in response to its container. 05

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