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 Answers to Chapter 13 Study Questions. 1. True: b), c) and d) 2. Reaction = 2 x reverse of original reaction. So $K_2 = (1/K_1)^2 = (1/0.447)^2 = 5.00$. 3. $K = [NH_3]^2 / [N_2] \times [H_2]^3$; more product made. Increasing the pressure increases the concentration; counteract this by shifting to the side with fewer moles.**

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13.1 Chemical Equilibria - Chemistry

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Answer/Explanation. Answer: d. Explanation: 4. An organic compound C 7 H 9 N forms clear solution when dissolved in KOH after reacting with C 6 H 5 SO 2 Cl, 'A' on diazotisation at 0°C and then reaction with ?-naphthol gives orangish red dye. 'A' on electrophilic substitution gives single product. 'A' is.

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